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(54) **PROCESS OF PREPARING A WATER DISPERSIBLE HYDROPHOBIC OR AEROPHILIC SOLID**  
**VERFAHREN ZUR HERSTELLUNG EINES WASSERDISPERGIERBAREN HYDROPHOBISCHEN ODER AEROPHILEN FESTKÖRPERS**  
**PROCEDE DE PREPARATION D'UN SOLIDE HYDROPHOBE OU AEROPHILE DISPERSIBLE DANS L'EAU**

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| <b>EP-A- 34 771</b>  | <b>EP-A- 0 074 050</b>  |                     |                        |                        |                        |                     |                      |                      |                        |                        |                        |                        |                        |                        |                        |
| <b>EP-A- 0 333 523</b>   | <b>EP-A- 0 347 751</b>  |                     |                        |                        |                        |                     |                      |                      |                        |                        |                        |                        |                        |                        |                        |
| <b>EP-B- 74 050</b>  | <b>CH-A- 420 817</b>  |                     |                        |                        |                        |                     |                      |                      |                        |                        |                        |                        |                        |                        |                        |
| <b>CH-A- 431 252</b>   | <b>DE-A- 1 617 737</b>  |                     |                        |                        |                        |                     |                      |                      |                        |                        |                        |                        |                        |                        |                        |
| <b>DE-A- 2 820 981</b>   | <b>GB-A- 2 122 085</b>  |                     |                        |                        |                        |                     |                      |                      |                        |                        |                        |                        |                        |                        |                        |
| <b>US-A- 2 876 160</b>   | <b>US-A- 3 529 065</b>  |                     |                        |                        |                        |                     |                      |                      |                        |                        |                        |                        |                        |                        |                        |
| <b>US-A- 4 006 025</b>   | <b>US-A- 4 726 955</b>  |                     |                        |                        |                        |                     |                      |                      |                        |                        |                        |                        |                        |                        |                        |
| (30) Priority: <b>02.11.1989 DK 546289</b>   | <ul style="list-style-type: none"><li>• <b>Synthesis and Characterisation of L-Ascorbate Phosphates and their Stabilities in Model Sytem, Kansas State University, Ph.D, 1976, pages 9 and 113;</b></li><li>• <b>Toxologisch-hygienische Beurteilung von Lebensmittelinhalts- und -zusatzstoffen sowie bedenklicher Verunreinigungen, Verlag Paul Parey, Berlin und Hamburg, 1987, Seiten 130, 136-139;</b></li><li>• <b>Merck Index, 10th edition, pages 258-259;</b></li><li>• <b>Development in Food Colours 1, pp 68/69;</b></li></ul>                  |                     |                        |                        |                        |                     |                      |                      |                        |                        |                        |                        |                        |                        |                        |
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| (73) Proprietor: <b>DANOCEMO A/S</b><br><b>DK-2750 Ballerup (DK)</b>   |   |                     |                        |                        |                        |                     |                      |                      |                        |                        |                        |                        |                        |                        |                        |
| (72) Inventors: <ul style="list-style-type: none"><li>• <b>MUSAEUS JENSEN, Nina</b><br/><b>DK-2900 Hellerup (DK)</b></li><li>• <b>VILSTRUP, Per</b><br/><b>DK-2200 Koebenhavn N (DK)</b></li><li>• <b>WINNING, Marianne</b><br/><b>DK-2980 Kokkedal (DK)</b></li></ul> |   |                     |                        |                        |                        |                     |                      |                      |                        |                        |                        |                        |                        |                        |                        |
| (74) Representative: <b>Kyed, Iver et al</b><br><b>c/o Hofman-Bang &amp; Boutard, Lehmann &amp; Ree A/S,</b><br><b>Hans Bekkevolds Allé 7</b><br><b>2900 Hellerup (DK)</b>   |   |                     |                        |                        |                        |                     |                      |                      |                        |                        |                        |                        |                        |                        |                        |

**EP 0 498 824 B2**

**Descripti n**

The present invention relates to a process of preparing a hydrophobic or aerophilic powdered solid which is dispersible in water or in an aqueous solution (in the following referred to as a water dispersible solid) in the form of discrete microparticles.

More specifically the invention relates to a process of preparing water dispersible hydrophobic or aerophilic powdered colorants and biologically active solids.

It is well known that the coloring effect of a colorant or pigment dispersed in an aqueous medium increases with decreasing particle size of the colorant or pigment and that the color intensity of the dispersion increases with increasing dispersability.

It is also well known that the bioavailability of biologically active powdered solids, such as carotenoids and drugs, such as griseofulvin, ibuprofen, benzodiazepines and hormones, such as progesteron dispersed in an aqueous medium increases with decreasing particle size of the dispersed solid

Therefore, there is a need for providing water dispersible hydrophobic or aerophilic solids of a small particle size. e.g. a particle size not exceeding 10  $\mu\text{m}$  and more preferably not exceeding 2  $\mu\text{m}$ .

Various methods of reducing the particle size of hydrophobic/aerophilic compounds to a maximum average of 10  $\mu\text{m}$  have been disclosed. It is known to mill  $\beta$ -carotene together with edible oil in a colloid mill (Chimia 21,329 (67)). Using this method it is possible to achieve a  $\beta$ -carotene content in the oil of about 30%.

Published Japanese patent application No. 57-3861 discloses a process for preparation of a water-dispersible carotenoid formulation which comprises mixing and grinding a carotenoid with gum arabic in a dry state using a grinding apparatus. Such as rotary ball mills, vibration ball mills, and hammer mills. Following grinding the solid product can be dissolved in water and the resulting solution can be converted into a powder by spray drying or freeze drying. A powder formulation having a carotenoid content of 1.5 percent by weight is obtained. A disadvantage of this method is that the milled compounds may be damaged due to the increase of temperature during grinding or milling in the dry state because of the lack of efficient cooling equipment. Furthermore, it is difficult to prevent oxidation of the milled compounds during a dry grinding or milling operation

Moreover, Chimia 21, 329 (1967). DE application No. 12 11 911 and DE publication No. 25 34 091 disclose methods of dissolving the active compound in a chlorinated organic solvent, emulsifying the solution in a gelatine/sucrose solution and expacting the solvent from the emulsion causing the active compound to crystallize in microcrystalline form. A disadvantage of this method is that it is technically impossible to remove the organic solvent completely and the solvent is a potential hazard during the process and as residues in the final product.

DK-B-154.395 discloses a preparation, which has been prepared by dissolving a carotenoid or a retinoid in an organic solvent that is miscible with water at a temperature of between 50 and 200°C optionally under pressure in less than 10 sec. The resulting molecular-disperse solution is immediately mixed with an aqueous solution of a hydrocolloid followed by an isolation of the colloid-disperse particles from the solvent to obtain a dry free flowing powder that can be dispersed in water. This method is disadvantageous because of the use of organic solvents that require special equipment for collection of the solvents. Furthermore, the concentration of the active compound will be relatively low in the final product, i.e. a maximum of about 20 percent by weight. Furthermore, it is stated that it has not been possible to mill hydrophobic solids, such as  $\beta$ -carotene in water or in an aqueous system to obtain the desired particle size without damage to the active compounds.

Research Disclosure (RD) 17064, June 1978, describes a method of preparing stable carotenoid colorants, wherein the carotenoids are milled in ball mills in an aqueous or oily medium in the presence of a protective colloid and a carrier, such as gelatine, and an ionic emulgator. The milling in the aqueous medium is followed by a spray drying process resulting in a water dispersible colorant having a carotenoid content of about 10%. However, the ionic emulgator will be present in the final product, which is undesirable especially when the colorant is to be used in food products.

It is well known that an aqueous dispersion of particles of hydrophobic/aerophilic solids can be prepared by mixing a hydrophobic/aerophilic solid with water in the presence of a wetting agent or tenside, such as lauryl sulfate or polysorbate. Without the use of a wetting agent milling is impossible, and without sufficient wetting the particles will agglomerate and the suspension will become too viscous for milling.

The object of the invention is to provide a process for preparing a hydrophobic/aerophilic solid that can be dispersed in water in the form of discrete microparticles without the use of oils, organic solvents and/or wetting agents or similar additives to form a microencapsulated product having a concentration of up to 71 % of hydrophobic/aerophilic solid.

This object can be achieved by the process of the invention which process comprises wetting the solid ..... (as in claim 1)

It is further assumed that the hydrocolloid, such as gelatine, provides sufficient wetting of the hydrophobic/aerophilic material to avoid agglomeraton of the fine particles formed during the milling process. The required minimum amount: of the hydrocolloid to provide sufficient wetting of the solid is dependent on the specific solid and hydrocolloid and the presence of other excipients.

It is assumed that the hydrocolloid protects the new surfaces resulting from the milling of the relatively coarse particles of the hydrophobic/aerophilic material

by creating a thin film that is intimately bound to the reactive new surfaces, thus preventing the particles to agglomerate due to hydrophobic interactions. Furthermore, the hydrocolloid reduces the surface tension and increases the dispersability of the final product.

Moreover, the hydrocolloid protects the milled particles in both the liquid and the dried state, thus preventing recrystallisation and crystal growth of materials.

Surprisingly, it has been found that by using the process of the invention microencapsulated products having a concentration of hydrophobic/aerophilic solid of up to 71% can be obtained.

The milling is preferably effected in a bead mill or any similar mill having cooling equipment for maintaining the temperature below a value at which the hydrophobic/aerophilic materials are decomposed and wherein the materials can be kept under a controlled atmosphere.

In a preferred embodiment of the process according to the invention the hydrophobic/aerophilic solid is preferably added to an aqueous solution of a hydrocolloid that has been degassed by boiling under vacuum and preferably covered by N<sub>2</sub> or any other inert gas. The mixture is then stirred at a temperature of between 0°C and 100°C, the lower limit being determined by the temperature at which the hydrocolloid forms a gel and the upper limit being determined by the heat lability of the active compound. Antioxidants may be added to the suspension.

The suspension is transferred to the mill which can be any type that is capable of milling the solid to a maximum particle size of 10µm, is equipped with a cooling jacket and wherein the atmosphere can be controlled. During the milling the temperature is between 0°C and 100°C, preferably above 60°C and the pressure is preferably between 0 and 5 bar. The suspension may be pumped through the mill in a peristaltic pump or using a gear pump having a flow of from 50 to 1000 ml/min, preferably from 150 to 400 ml/min.

Solid hydrophobic/aerophilic materials that can be milled and encapsulated in the process according to the invention include carotenoids, such as β-carotene, annatto, bixin, norbixin, capsanthin, capsorubin, lycopene, β-apo-8'-carotenal, flavoxanthin, lutein, cryptoxanthin, rubixanthin, violaxanthin, rhodoxanthin, canthaxanthin, astaxanthin and citranaxanthin and derivatives thereof: natural colorants, such as curcumin, chlorophyll, carmine, etc.;

The aqueous medium can optionally further contain excipients in an amount of up to 70 percent by weight of the suspension, such as a dissolved carbohydrate, such as sorbitol and sucrose, and/or an antioxidant or an oil containing an antioxidant.

The resulting suspension is finely divided and dried using any combination of conventional methods, such as spray cooling, spray drying, modified spray drying or sheet drying and crushing, etc.

Another advantage of the process according to the

invention is that the hydrocolloid used in the milling of a hydrophobic/aerophilic solid can act as a matrix material in a subsequent encapsulation process. Such processes will be described in the following.

#### Spray cooling

In a spray cooling process a suspension containing a hydrocolloid is preferably sprayed using an atomizing nozzle or an atomizing wheel at a temperature higher than the gelling/melting point, i.e. from about 38° to about 95°C and at a viscosity of preferably between 50 and 300 mPa.s in a spraying chamber, wherein the temperature is from 0° to about 40°C, thereby forming microcapsules of gelatinized hydrocolloid.

A powdery spraying excipient is preferably blown into the spraying chamber in order to prevent agglomeration of the gelatinized microcapsules and to prevent adherence to the chamber wall. The spraying excipient is preferably supplied in an amount of from 5 to 50 percent by weight based on the weight of the final product.

The microcapsules are transferred to a fluidized bed, wherein they may be dried to a residual water content of between 0 and 10% (preferably from 2 to 5%) and in which excessive spraying excipient is separated. The drying air temperature is preferably from about 0° to about 60°C.

#### Modified spray drying

The modified spray drying process differs from the spray cooling process in that the temperature in the spraying chamber is high, preferably between 50° and 95°C.

In the modified spray drying process the suspension is preferably sprayed at a temperature of from 5° to 99°C and at a viscosity of from 50 to 300 mPa.s using an atomizing nozzle or an atomizing wheel in a spraying chamber, wherein the temperature is from 50° to 95°C.

A powdery spraying excipient may be blown into the spraying chamber in order to prevent agglomeration of the formed microcapsules and to prevent adherence to the chamber wall. The spraying excipient is preferably supplied in an amount of from 5 to 50 percent by weight based on the weight of the final product.

The powdered microcapsules may be transferred to a fluidized bed, wherein they may be dried to a residual water content of between 0 and 10% (preferably from 2 to 5%) and excessive spraying excipient is separated. The drying air temperature is preferable from about 0° to about 60°C.

In the spray cooling, spray drying and modified spray drying processes the following spraying excipients may be used: starches, modified starches, tricalcium phosphate, lactose, mannitol, ethyl cellulose, coagulated albumin, hardened gelatine, casein, stearat-Ca, stearat-Na, metal soaps, hydrogenated coconut oil, polyoxide, talcum, waxes, and silicates.

In a sheet drying process the suspension is preferably dried in a thin layer to form a solidified suspension which subsequently may be ground into a powder. Alternatively, the suspension may be emulsified in an oil, washed and dried/spray dried or extruded, drum dried and crushed or treated by any combination of known methods of finely dividing and drying or drying and finely dividing.

Microencapsulated  $\beta$ -carotene may be a constituent of vitamin tablets. However, because of the relatively low concentration of  $\beta$ -carotene in microcapsules prepared using the known techniques, it is necessary to formulate the vitamin tablets with a relatively large amount of  $\beta$ -carotene microcapsules resulting in bulky tablets. This obstacle is overcome when tablets are formulated with microcapsules having a high  $\beta$ -carotene content that are prepared according to the invention.

Another advantage of the hydrocolloid protection of microparticles of hydrophobic/aerophilic compounds, is an improved performance when preparing tablets from such compounds.

It is preferred to include an antioxidant in the suspension when the active compound is sensitive to oxygen. The antioxidant can be water soluble or water insoluble and can be incorporated during the milling process or in a subsequent emulsification process.

The dispersability of the microencapsulated product is evaluated visually after the addition of 0.2 g of microencapsulated product to 200 ml of water having a temperature of 40°-45°C in a beaker. The dispersability is satisfactory when the microencapsulated particles are immediately wetted and a uniform dispersion is formed after 2 minutes of stirring. When viewed in a light microscope in transparency the dispersion consists of predominantly discrete particles.

In the microencapsulated product prepared according to the above process containing a hydrophobic/aerophilic solid having a maximum average particle size not exceeding 10  $\mu\text{m}$  and which can be dispersed in water in the form of discrete microparticles, the amount of hydrophobic/aerophilic solid is up to 71 percent by weight of the microencapsulated product. The microcapsules prepared by the process described above can be used in pharmaceutical compositions, feeds and foodstuffs.

#### Example 1

1000 g of  $\beta$ -carotene was added in a  $\text{N}_2$ -atmosphere to a solution of 584 g of 240 Bloom gelatine and 100 g of Na-ascorbate in 2800 g of water having a temperature of 65°C. Before the addition of  $\beta$ -carotene the solution had been degassed using a vacuum and the pressure had been brought to equilibrium with  $\text{N}_2$ . When the  $\beta$ -carotene crystals were sufficiently wetted, the suspension was milled in a bead mill, DynoMill type KDL, during 1 hour of recirculation. The milled suspension was transferred in a  $\text{N}_2$ -atmosphere to a degassed

aqueous solution of 1300 g of gelatine and 2044 g of sucrose, wherein 9.5 g of ascorbyl palmitate and 14.3 g of a mixture of  $\alpha$ -,  $\beta$ -, and  $\gamma$ -tocopherol had been emulsified. After being thoroughly mixed the resulting suspension was spray cooled in a conventional manner.

The following intervals of product characteristics for four products were obtained.

$\beta$ -carotene content: 13.2 - 13.8 percent by weight  
extractable  $\beta$ -carotene: 0.3 - 0.8 percent by weight  
absorption ratio  $A_{452\text{nm}}:A_{483\text{nm}}$ : 1.16  
absorption ratio  $A_{452\text{nm}}:A_{340\text{nm}}$ : 14.7 - 15.2  
stability after 6 months at 22°C 100% of original  $\beta$ -carotene content

The visual dispersability test was satisfactory.

#### Example 2

600 g of  $\beta$ -carotene was added in a  $\text{N}_2$ -atmosphere to a solution of 350 g of 240 Bloom gelatine and 60 g of Na-ascorbate in 1500 g of water having a temperature of 65°C. Before the addition of  $\beta$ -carotene the solution had been degassed using a vacuum and the pressure had been brought to equilibrium with  $\text{N}_2$ . When the  $\beta$ -carotene crystals were sufficiently wetted, the suspension was milled in a bead mill, DynoMill type KDL, during 1 hour of recirculation. The milled suspension was transferred to a degassed aqueous solution of 184 g of gelatine and 484 g of sucrose in a  $\text{N}_2$ -atmosphere, wherein 7.9 g of a mixture of  $\alpha$ -,  $\beta$ -, and  $\gamma$ -tocopherol had been emulsified. After being thoroughly mixed the resulting suspension was spray cooled in a conventional manner.

The following product characteristics were obtained:

$\beta$ -carotene content: 16.9 percent by weight  
extractable  $\beta$ -carotene: 0.4 percent by weight  
absorption ratio  $A_{452\text{nm}}:A_{483\text{nm}}$ : 1.16  
absorption ratio  $A_{452\text{nm}}:A_{340\text{nm}}$ : 15.4  
stability after 3 months at 22°C: 98.7% of original  $\beta$ -carotene content

The visual dispersability test was satisfactory.

#### Example 3

500 g of  $\beta$ -carotene was added to a solution of 146 g of 240 Bloom gelatine and 50 g of Na-ascorbate in 1000 g of water having a temperature of 65°C. When the  $\beta$ -carotene crystals were sufficiently wetted, the suspension was milled in a bead mill, DynoMill type KDL, during 1 hour of recirculation. In the milled suspension 7.1 g of a mixture of  $\alpha$ -,  $\beta$ -, and  $\gamma$ -tocopherol had been emulsified. After being thoroughly mixed the resulting suspension was spray cooled in a conventional manner.

The following product characteristics were

obtained:

$\beta$ -carotene content: 42.5 percent by weight  
absorption ratio  $A_{452nm}:A_{483nm}$ : 1.19  
absorption ratio  $A_{452nm}:A_{340nm}$ : 10.3

The visual dispersability test was satisfactory.

#### Example 4

360 g of  $\beta$ -carotene was added to a solution of 210 g of low Bloom gelatine and 36 g of Na-ascorbat in 935 g of water having a temperature of 65°C. When the  $\beta$ -carotene crystals were sufficiently wetted, the suspension was milled in a bead mill, DynoMill type KDL, during 1 hour of recirculation. The milled suspension was transferred to an aqueous solution of 648 g of high Bloom gelatine and 556 g of sucrose, wherein a mixture of 151 g of coconut oil and 0.9 g of a mixture of  $\alpha$ -,  $\beta$ - and  $\gamma$ -tocopherol had been emulsified. After being thoroughly mixed the resulting suspension was spray cooled in a conventional manner.

The following product characteristics were obtained:

$\beta$ -carotene content: 11.3 percent by weight  
extractable  $\beta$ -carotene: 0.1 percent by weight  
absorption ratio  $A_{452nm}:A_{483nm}$ : 1.16  
absorption ratio  $A_{452nm}:A_{340nm}$ : 14.8  
stability after 3 months at 22°C: 96.5% of original  $\beta$ -carotene content

The visual dispersability test was satisfactory.

#### Example 5

600 g of  $\beta$ -carotene was added in a  $N_2$ -atmosphere to a solution of 350 g of 70 Bloom gelatine and 60 g of Na-ascorbat in 1500 g of water having a temperature of 65°C. Before the addition of  $\beta$ -carotene the solution had been degassed using a vacuum and the pressure had been brought to equilibrium with  $N_2$ . When the  $\beta$ -carotene crystals were sufficiently wetted, the suspension was milled in a bead mill, DynoMill type KDL, during 1 hour of recirculation. The milled suspension was transferred to a degassed aqueous solution of 592 g of 30 Bloom gelatine and 558 g of sucrose in a  $N_2$ -atmosphere, wherein a mixture of 117 g of coconut oil and 0.7 g of a mixture of  $\alpha$ -,  $\beta$ -, and  $\gamma$ -tocopherol had been emulsified. After being thoroughly mixed the resulting suspension was subjected to a modified spray drying.

The following product characteristics were obtained:

$\beta$ -carotene content: 11.4 percent by weight  
extractable  $\beta$ -carotene: 0.7 percent by weight  
absorption ratio  $A_{452nm}:A_{483nm}$ : 1.16  
absorption ratio  $A_{452nm}:A_{340nm}$ : 14.4

stability after 3 months at 22°C: 95.1% of original  $\beta$ -carotene content

The visual dispersability test was satisfactory.

#### Example 6

25 g of Canthaxanthin was added to a solution of 14.5 g of 106 Bloom gelatine and 2.5 g of Na-ascorbat in 70 g of water having a temperature of 65°C. When the cantaxanthin crystals were sufficiently wetted, the suspension was milled in a bead mill, DynoMill type KDL, during 1 hour. The milled suspension was transferred to an aqueous solution of 153 g of 240 Bloom gelatine and 153 g of sucrose, wherein 0.5 g of ascorbyl palmitate and 2.0 g of BHT and 10.0 g of ethoxyquin had been emulsified. After being thoroughly mixed the resulting suspension was spray cooled in a conventional manner.

The final product contained 4.4 percent by weight of canthaxanthin.

The visual dispersability test was satisfactory.

#### Example 7

25 g of Astaxanthin was added to a solution of 14.5 g of 106 Bloom gelatine and 2.5 g of Na-ascorbat in 1500 g of water having a temperature of 65°C. When the astaxanthin crystals were sufficiently wetted, the suspension was milled in a bead mill, DynoMill type KDL, during 1 hour. The milled suspension was transferred to an aqueous solution of 153 g of 240 Bloom gelatine and 153 g of sucrose, wherein 0.5 g of ascorbyl palmitate and 2.0 g of BHT and 10.0 g of ethoxyquin had been emulsified. After being thoroughly mixed the resulting suspension was spray cooled in a conventional manner.

The final product contained 4.5 percent by weight of astaxanthin.

The visual dispersability test was satisfactory.

#### Example 8

500 g of  $\beta$ -carotene was added to a solution of 250 g of gum arabic and 50 g of Na-ascorbat in 1750 g of water having a temperature of 65°C. When the  $\beta$ -carotene crystals were sufficiently wetted, the suspension was milled in a bead mill, DynoMill type KDL, during 1 hour of recirculation. The milled suspension was transferred to an aqueous solution of 916 g of gum arabic and 780 g of sucrose, wherein 7.4 g of a mixture of  $\alpha$ -,  $\beta$ - and  $\gamma$ -tocopherol had been emulsified. After being thoroughly mixed the resulting suspension was spray cooled in a conventional manner.

The following product characteristics were obtained:

$\beta$ -carotene content: 8.9 percent by weight  
absorption ratio  $A_{452nm}:A_{483nm}$ : 1.16

absorption ration  $A_{452nm}:A_{340nm}$ : 14.2

The visual dispersability test was satisfactory

#### Comparative example

500 g of  $\beta$ -carotene was added to a solution of 200 g of Methocel<sup>®</sup> ES and 50 g of Na-ascorbat in 2000 g of water having a temperature of 65°C. When the  $\beta$ -carotene crystals were sufficiently wetted, the suspension was milled in a bead mill, DynoMill type KDL, during 1 hour of recirculation. The milled suspension was transferred to an aqueous solution of 323 g of Methocel<sup>®</sup> E5 and 1224 g of sucrose, wherein 7.6 g of a mixture of  $\alpha$ -,  $\beta$ -, and  $\gamma$ -tocopherol had been emulsified. After being thoroughly mixed the resulting suspension was subjected to a modified spray drying process in a conventional manner.

The following product characteristics were obtained:

$\beta$ -carotene content: 4.0 percent by weight  
absorption ration  $A_{452nm}:A_{483nm}$ : 1.19  
absorption ration  $A_{452nm}:A_{340nm}$ : 10.0

The visual dispersability test was satisfactory

#### Example 9

500 g of  $\beta$ -carotene was added to a solution of 500 g of Capsul<sup>®</sup> 50 g and of Na-ascorbat in 1500 g of water having a temperature of 65°C. When the  $\beta$ -carotene crystals were sufficiently wetted, the suspension was milled in a bead mill, DynoMill type KDL, during 1 hour of recirculation. The milled suspension was transferred to an aqueous solution of 665 g of Capsule and 777 g of sucrose, wherein 7.2 g of a mixture of  $\alpha$ -,  $\beta$ -, and  $\gamma$ -tocopherol had been emulsified. After being thoroughly mixed the resulting suspension was subjected to a modified spray drying process in a conventional manner.

The following product characteristics were obtained:

$\beta$ -carotene content: 9.9 percent by weight  
absorption ration  $A_{452nm}:A_{483nm}$ : 1.16  
absorption ration  $A_{452nm}:A_{340nm}$ : 13.6

The visual dispersability test was satisfactory.

#### Example 10

500 g of  $\beta$ -carotene was added to a solution of 292 g of 240 Bloom gelatine in 1400 g of water having a temperature of 65°C. When the  $\beta$ -carotene crystals were sufficiently wetted, the suspension was milled in a bead mill. DynoMill type KDL, during 1 hour of recirculation. The milled suspension was transferred to an aqueous solution of 1132 g of 240 Bloom gelatine and 1298 g of

sucrose, wherein 7.4 g of a mixture of  $\alpha$ -,  $\beta$ -, and  $\gamma$ -tocopherol had been emulsified. After being thoroughly mixed the resulting suspension was spray cooled in a conventional manner.

The following product characteristics were obtained:

$\beta$ -carotene content: 10.5 percent by weight  
extractable  $\beta$ -carotene: 0.3 percent by weight  
absorption ration  $A_{452nm}:A_{483nm}$ : 1.17  
absorption ration  $A_{452nm}:A_{340nm}$ : 14.6

The visual dispersability test was satisfactory.

#### Example 11

300 g of Curcumin crystals was added to a solution of 300 g of low-bloom gelatine and, 300 g of sucrose in 900 g of water having a temperature of 65°C. When the curcumin crystals were sufficiently wetted, the suspension was milled in a bead mill, DynoMill type KDL, during 1 hour of recirculation. The milled suspension was transferred to a solution of 750 g of sucrose in 300 g of water.

After being thoroughly mixed the resulting suspension was subjected to a modified spray drying process.

The final product contained 26.0 percent by weight of curcumin.

#### Claims

1. A process for preparing a hydrophobic/aerophilic solid selected from the group consisting of carotenoids and derivatives thereof natural colorants and which solid is dispersable in water in the form of discrete microparticles comprising wetting the solid in an aqueous medium solely by means of a hydrocolloid selected from the group consisting of gelatine, gum arabic, soya bean protein and/or modified starch in an amount of not less than about 10 percent by weight of the solid, and milling the wetted solid to obtain a suspension containing suspended particles of an average particle size not exceeding 10  $\mu$ m and finely dividing and drying the suspension thus formed to obtain a powder.
2. A process according to claim 1, characterized in that the suspension containing suspended particles is finely divided and dried by spray cooling, conventional spray drying or modified spray drying.
3. A process according to claim 2, characterized in that the suspension is finely divided and dried in the presence of a spraying excipient.
4. A process according to any of the preceding claims, characterized in that the carotenoid is  $\beta$ -carotene,  $\beta$ -apo-8'-carotenal, canthaxanthin, and/or astaxanthin.

thin.

5. A process according to any of claims 1-3, **characterized** in that the natural colorant is curcumin and/or chlorophyll.

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6. A process according to any one of the preceding claims, **characterized** in that the suspension contains a carbohydrate, in an amount of up to 70 percent by weight of the suspension.

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7. A process according to claim 6, **characterized** in that the carbohydrate is sucrose.

8. A process according to any one of the preceding claims, **characterized** in that the suspension further contains an antioxidant.

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9. A process according to claim 1, **characterized** in that the solid is used in an amount which is sufficient to provide a microencapsulated product having a concentration of up to 71 percent by weight of solid active material.

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#### Patentansprüche

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1. Verfahren zur Herstellung eines hydrophoben/aerophilen Feststoffes, ausgewählt aus der aus Carotinoiden und Derivaten hiervon und natürlichen Farbstoffen bestehenden Gruppe, und wobei dieser Feststoff in Wasser in Form von diskreten Mikroteilchen dispergierbar ist, umfassend das Benetzen dieses Feststoffes in einen wässrigen Medium nur mittels eines Hydrokolloids, ausgewählt aus der aus Gelatine, Gummi arabicum, Sojabohnenprotein und/oder modifizierter Stärke bestehenden Gruppe, in einer Menge von nicht weniger als etwa 10 Gew.-% des Feststoffes, und Mahlen des benetzten Feststoffes zum Erhalt einer Suspension, die suspendierte Teilchen mit einer 10 µm nicht übersteigenden Durchschnittsteilchengröße enthält, und Feinunterteilen und Trocknen der so gebildeten Suspension zum Erhalt eines Pulvers.

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2. Verfahren nach Anspruch 1, dadurch gekennzeichnet, daß die Suspension, die suspendierte Teilchen enthält, durch Sprühkühlung, konventionelles Sprühtrocknen oder modifiziertes Sprühtrocknen feinunterteilt wird.

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3. Verfahren nach Anspruch 2, dadurch gekennzeichnet, daß die Suspension in Anwesenheit eines Sprühtrocknmittels feinunterteilt und getrocknet wird.

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4. Verfahren nach einem der vorhergehenden Ansprüche, dadurch gekennzeichnet, daß das Carotinoid β-Carotin, β-apo-8'-Carotinal, Canthaxanthin

und/oder Astaxanthin ist.

5. Verfahren nach Anspruch 1 bis 3, dadurch gekennzeichnet, daß der natürliche Farbstoff Curcumin und/oder Chlorophyll ist.

6. Verfahren nach einem der vorhergehenden Ansprüche, dadurch gekennzeichnet, daß die Suspension ein Kohlenhydrat in einer Menge bis zu 70 Gew.-% der Suspension enthält.

7. Verfahren nach Anspruch 11, dadurch gekennzeichnet, daß das Kohlenhydrat Saccharose ist.

8. Verfahren nach eine der vorhergehenden Ansprüche, dadurch gekennzeichnet, daß die Suspension weiterhin ein Antioxidans enthält.

9. Verfahren nach Anspruch 1, dadurch gekennzeichnet, daß der Feststoff in einer Menge eingesetzt wird, die ausreicht, um ein mikroeingekapseltes Produkt zu liefern, das eine Konzentration von bis zu 71 Gew.-% an feste aktivem Material hat.

#### Revendications

1. Procédé de préparation d'un solide hydrophobe/aérophile choisi dans le groupe formé par les caroténoïdes et leurs dérivés et les colorants naturels, ce solide pouvant être dispersé dans l'eau sous forme de microparticules séparées, comprenant le mouillage du solide par un milieu aqueux uniquement à l'aide d'un hydrocolloïde choisi dans le groupe formé par la gélatine, la gomme arabique, la protéine de soja et/ou l'amidon modifié, en quantité qui n'est pas inférieure à 10 % du poids du solide, et le broyage du solide mouillé pour l'obtention d'une suspension qui contient des particules en suspension dont la dimension particulaire moyenne ne dépasse pas 10 µm, et la division fine et le séchage de la suspension ainsi formée pour l'obtention d'une poudre.

2. Procédé selon la revendication 1, caractérisé en ce que la suspension qui contient des particules est finement divisée et séchée par refroidissement avec pulvérisation, par séchage traditionnel par atomisation ou par séchage par atomisation modifié.

3. Procédé selon la revendication 2, caractérisé en ce que la suspension est finement divisée et séchée en présence d'un excipient de pulvérisation.

4. Procédé selon l'une quelconque des revendications précédentes caractérisé en ce que le caroténoïde est du β-carotène, du β-apo-8'-caroténal, de la

cathaxanthine et/ou de l'astaxanthine.

5. Procédé selon l'une quelconque des revendications 1 à 3, caractérisé en ce que le colorant naturel est la curcumine et/ou la chlorophylle. 5
6. Procédé selon l'une quelconque des revendications précédentes, caractérisé en ce que la suspension contient un glucide en quantité pouvant atteindre 70 % du poids de la suspension. 10
7. Procédé selon la revendication 6, caractérisé en ce que le glucide est le saccharose.
8. Procédé selon l'une quelconque des revendications précédentes, caractérisé en ce que la suspension contient en outre un antioxydant. 15
9. Procédé selon la revendication 1, caractérisé en ce que le solide est utilisé en quantité qui suffit pour former un produit microencapsulé ayant une concentration pouvant atteindre 71 % en poids de matière active solide. 20

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